

What can we do for you?

Focus areas: By technique

Separations

- Chromatography (LC)
- Field Flow Fractionation (FFF)
- CRYSTAF, TREF, CEF

Spectroscopy

- Infrared
- Raman
- NMR

Thermal

- DSC
- TGA
- Pyrolysis GC-MS

Imaging

- Infrared-microscopy
- Raman-microscopy
- Polarized light microscopy
- Electron Microscopy

Seite 1

THE POWER OF LIQUID CHROMATOGRAPHY

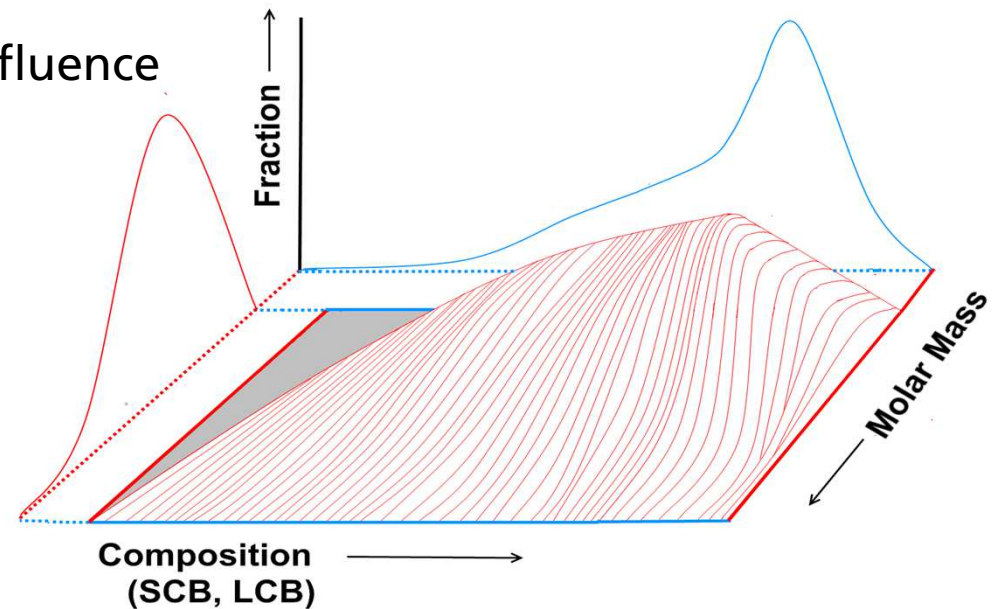


Photo: Fraunhofer

Challenges in polymer analysis

Multivariate molecular distribution

- Polymers show distributions in various molecular metrics e.g.: molar mass, architecture, end-capping
- Complex molecular properties influence macroscopic behavior
- ➔ Polymer represent highly complex materials
- ➔ No characterization techniques that allow a simultaneous and comprehensive analysis of all molecular metrics

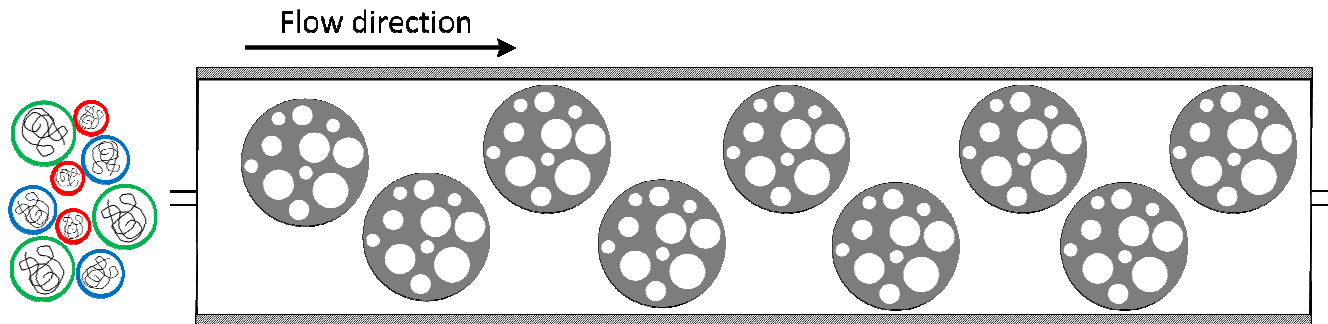
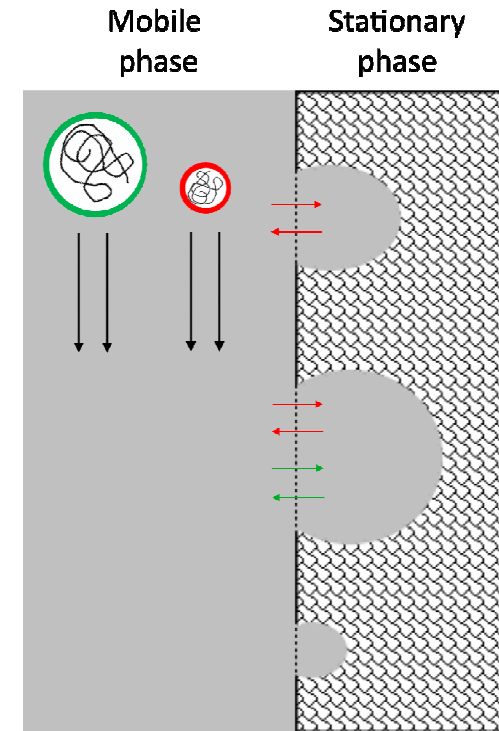


Size-exclusion chromatography

Basic separation principle

- Size-exclusion effect: macromolecules permeate in pores of the stationary phase
- Larger molecules are partially excluded from pores
- ➔ Inverse sieve effect based on entropic processes

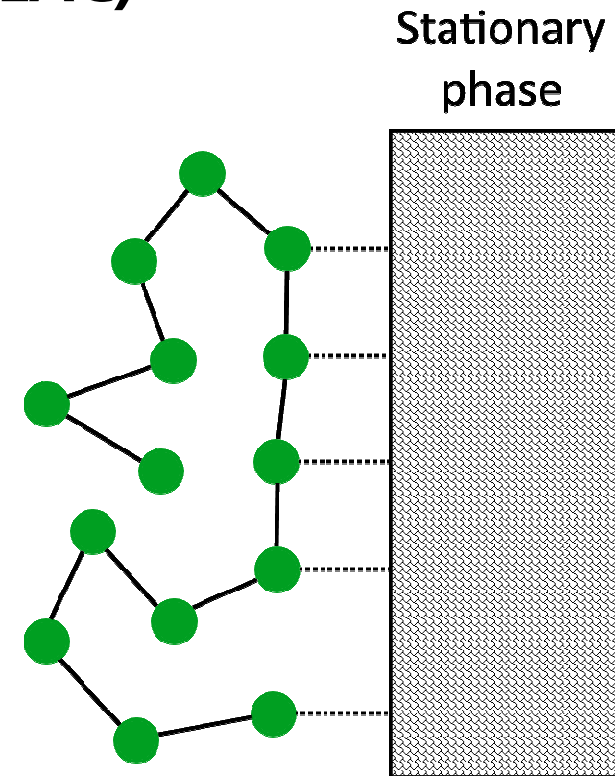
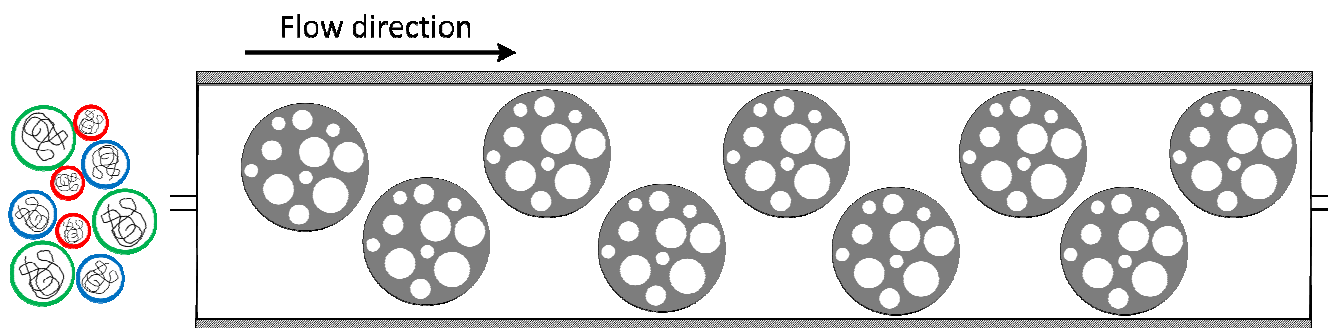
 molar masses \uparrow \rightarrow elution volume \downarrow



Liquid adsorption chromatography (LAC)

Basic separation principle

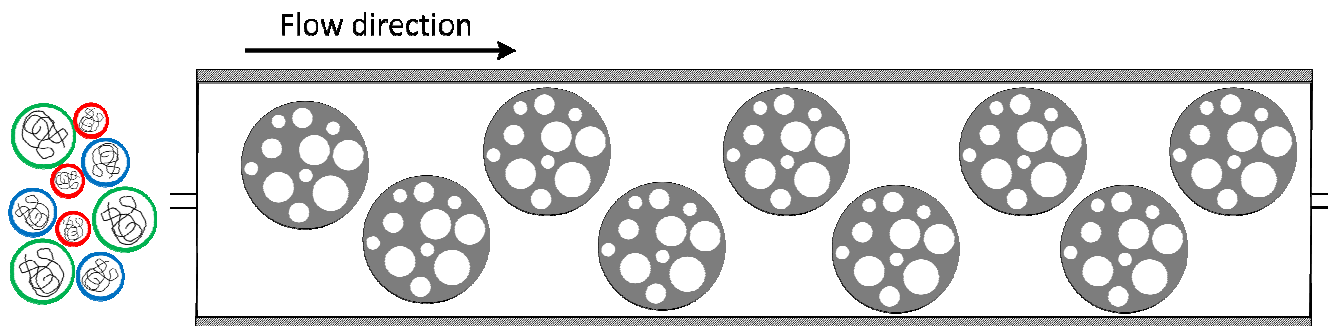
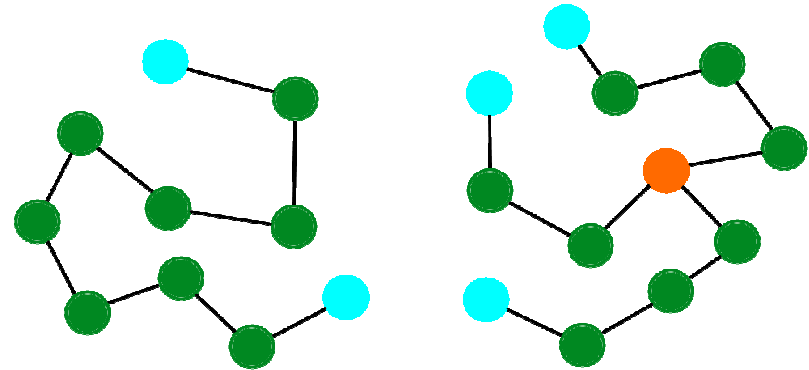
- Each repeating unit can interact with column
- molar masses \uparrow \rightarrow elution volume \uparrow
- Separation according to molar mass and chemical composition



Liquid chromatography under critical conditions

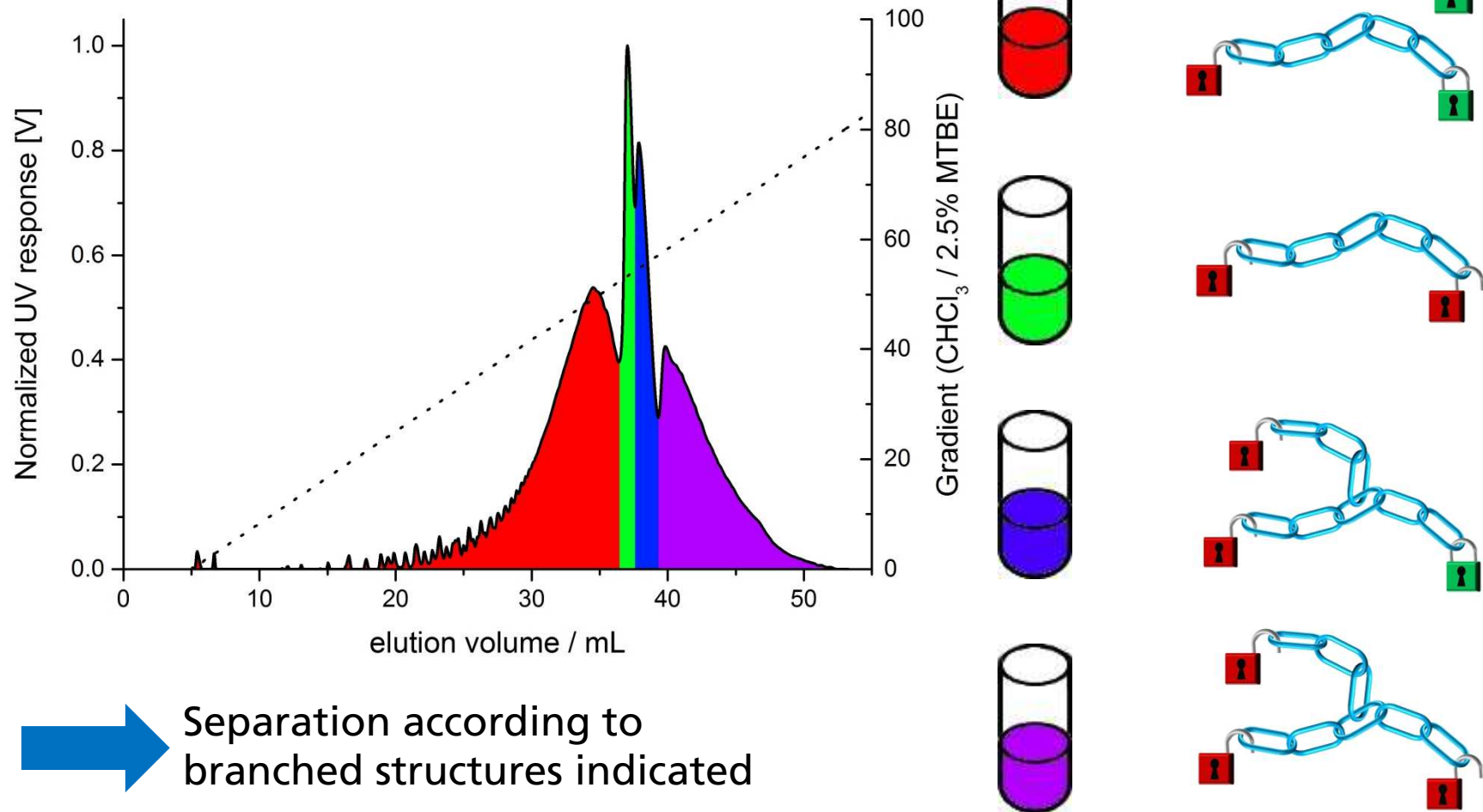
Basic separation principle

- Size-exclusion effects and interactions are balanced out
- Polymer backbone becomes “invisible”
- Separation according to end-groups or branching units is possible
- Separation is possible according to branching based on end-group moiety or branching units of the polymer



Solvent gradient at near-critical conditions (SG-NCC)

Fractionation



Two-dimensional liquid chromatography (2D-LC)

- 1D: Solvent gradient at near-critical conditions (SG-NCC)
- 2D: Size-exclusion chromatography (SEC)
- Improvement in separation: **chemical structure** and **molar mass**
- ➔ Comprehensive analysis of branched structures

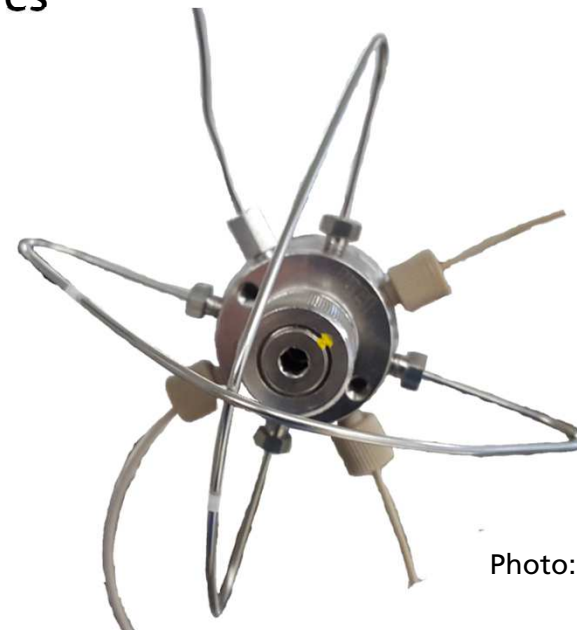
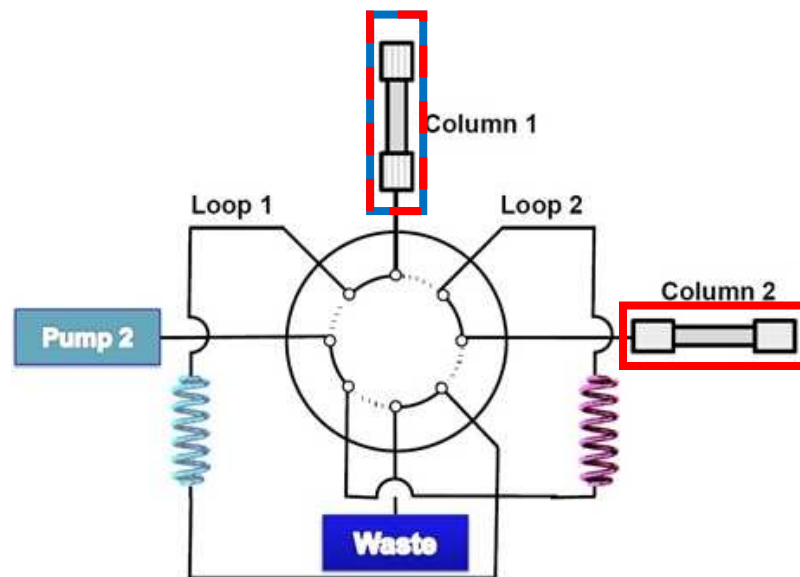
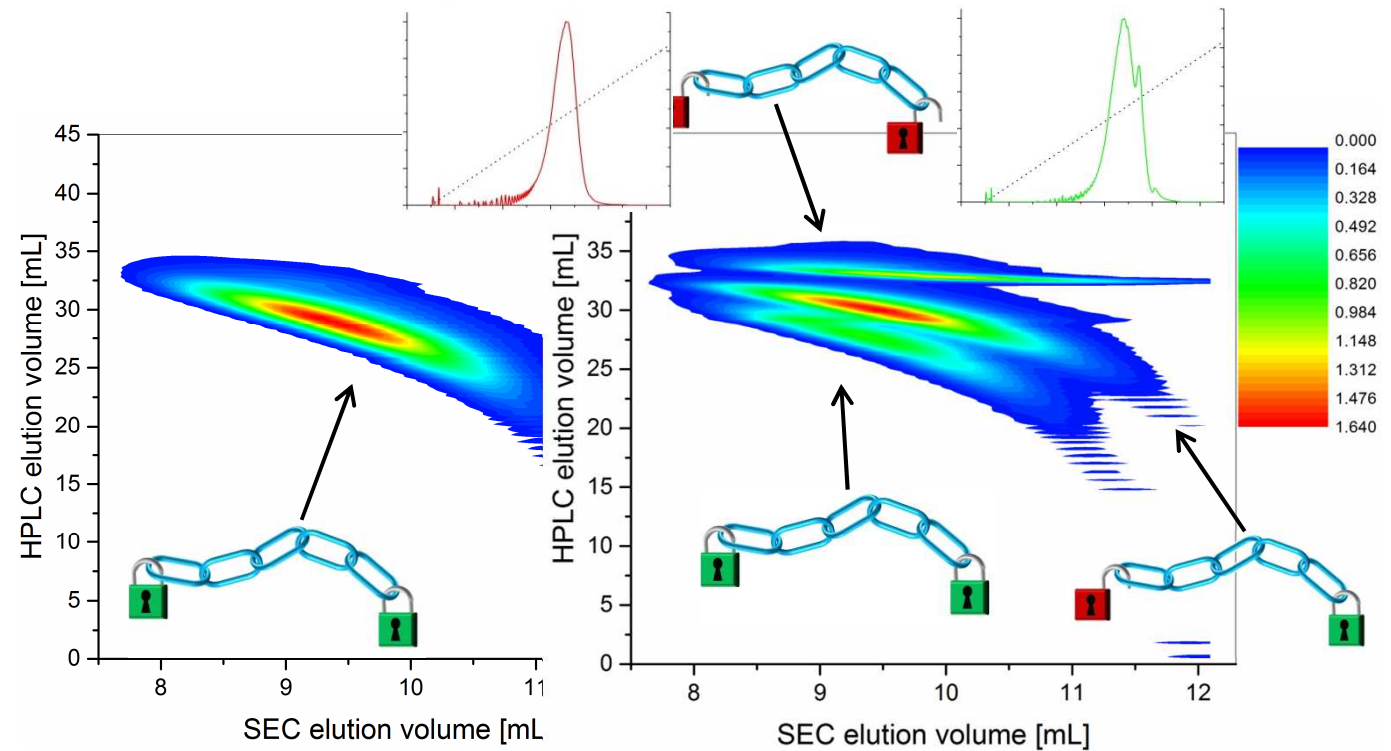


Photo: N. Apel

Two-dimensional liquid chromatography (2D-LC)

End-capped + linear

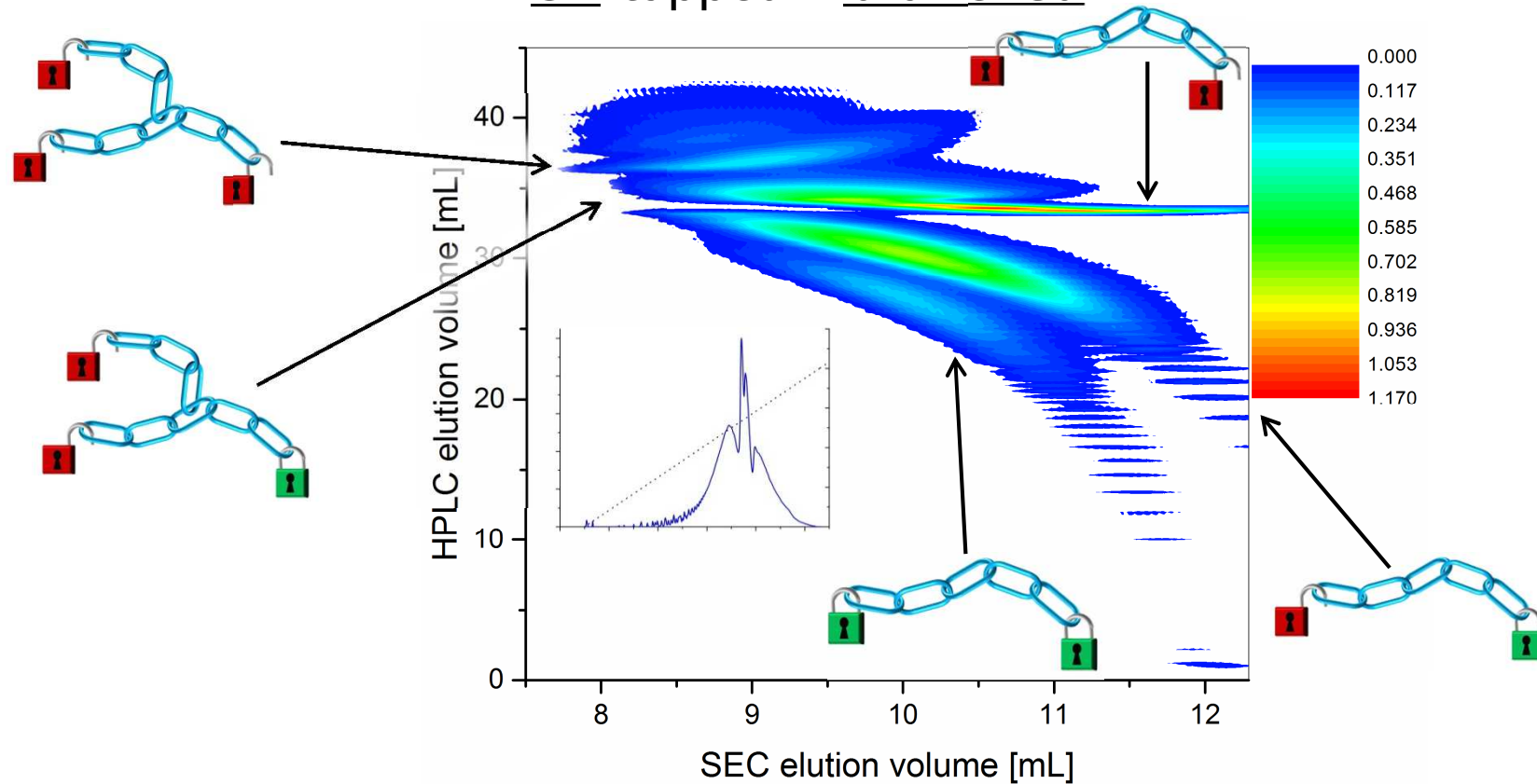
Un-capped + linear



➡ Complete separation according to end-groups

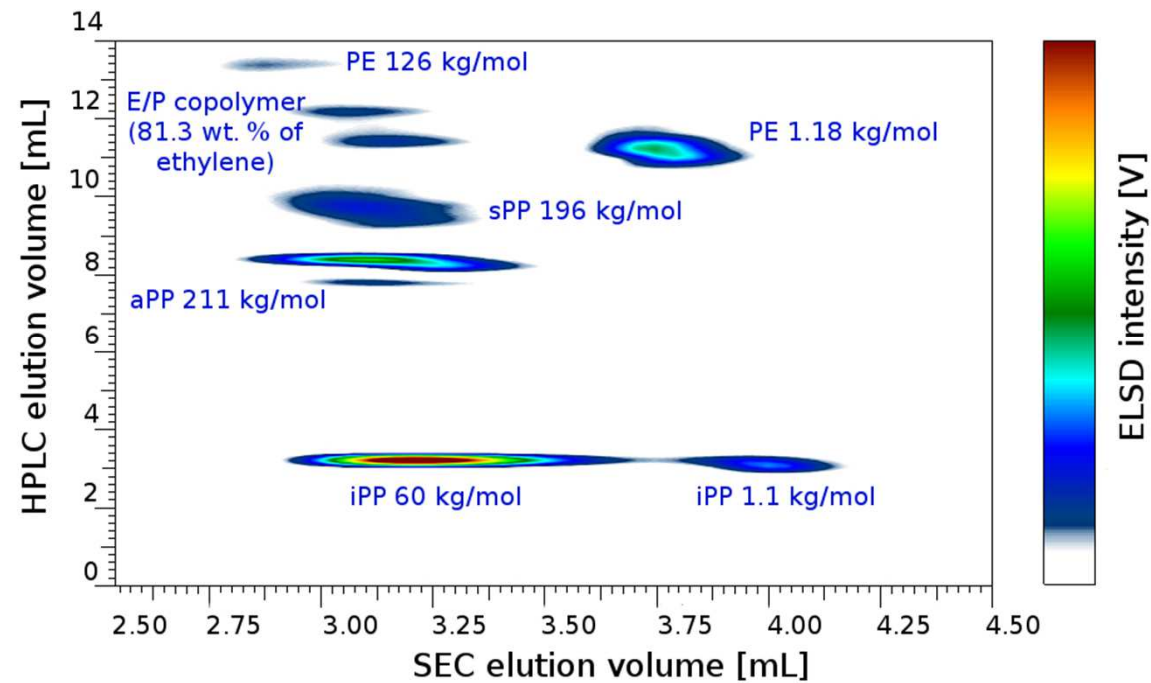
Two-dimensional liquid chromatography (2D-LC)

Un-capped + branched



➡ Separation according to branched structures

HT 2D-LC of polyolefins



HPLC: Hypercarb™, 1-decanol → TCB, 0.1 mL/min, 140 °C

SEC: PL Rapide™ H, TCB, 2.5 mL/min, 160 °C

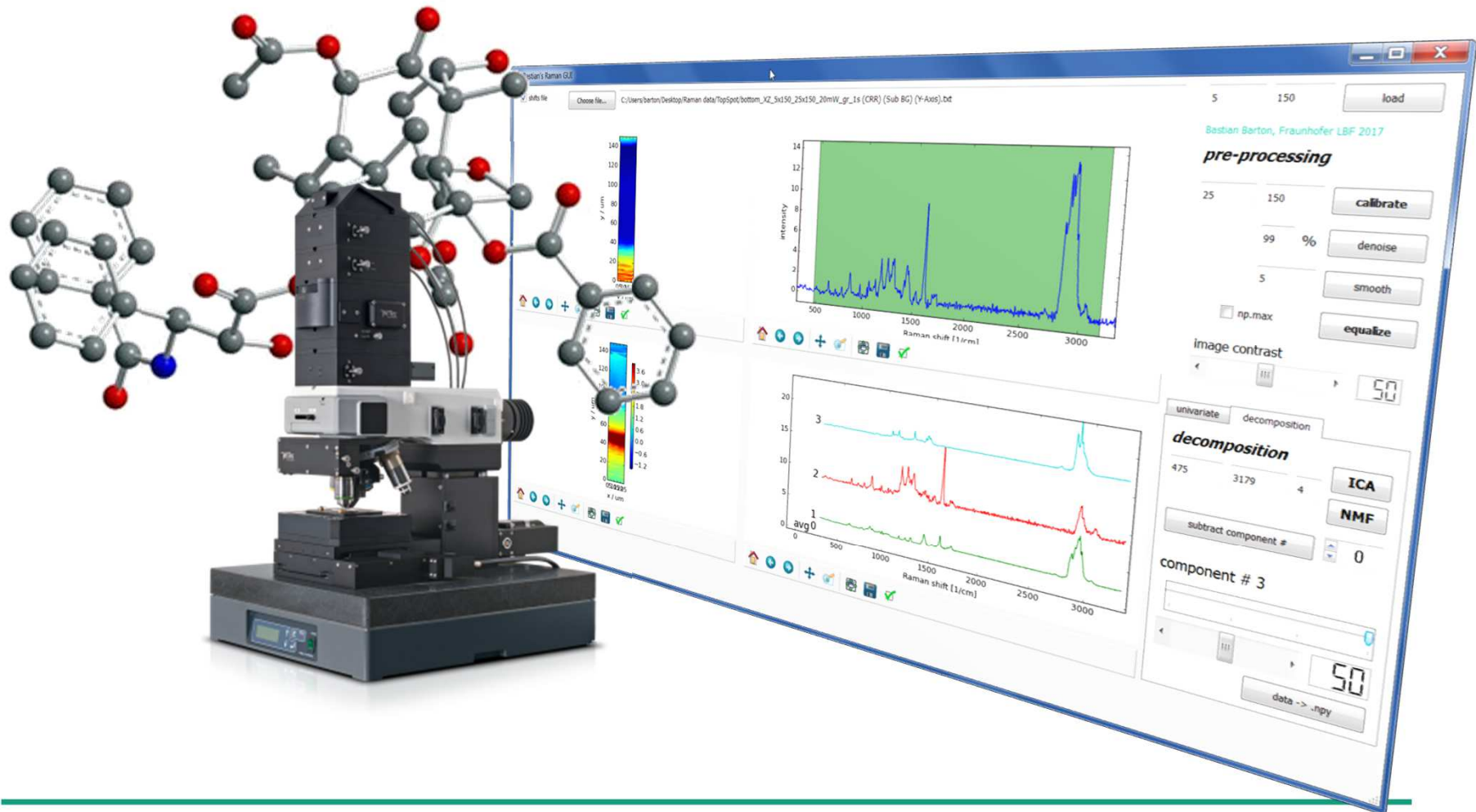
A. Ginzburg, T. Macko, V. Dolle, R. Brüll, *J. Chromatog. A* **2010**, 1217, 6867

A. Ginzburg, T. Macko, V. Dolle, R. Brüll, *Eur. Polym. J.*, **2011**, 47, 319

What liquid chromatography can do for you:

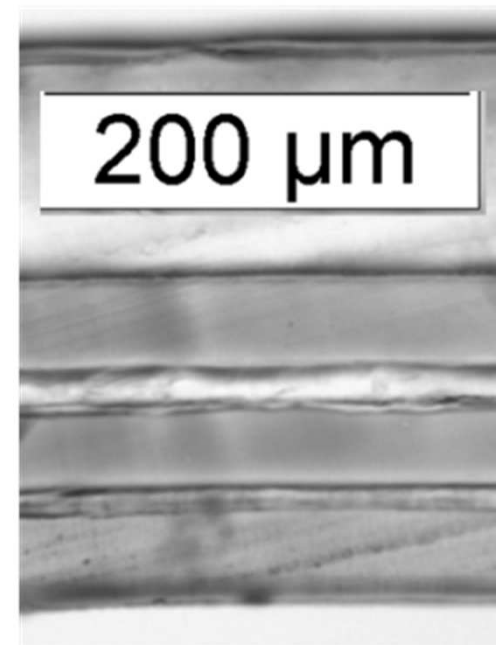
- TD-SEC: Monitoring of branching along the molar mass distribution
- SG-NCC: Separation according to end-capping level and branching
- 2D-LC: Individually separated linear and branched end-group species
- 2D-LC with TD-SEC: Detailed determination on branching parameters for individual branched species

HIGH RESOLUTION CHEMICAL MICROSCOPY OF INDUSTRIAL POLYMERS

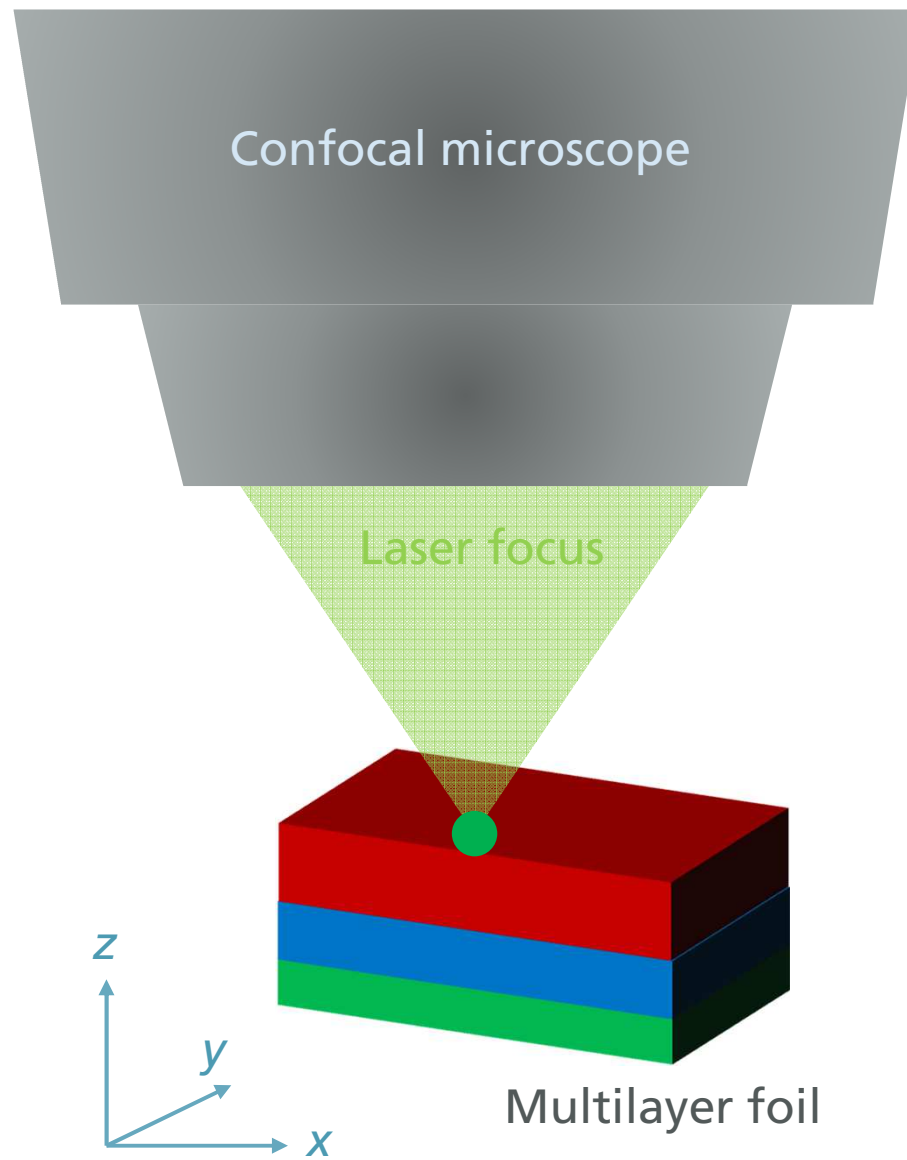


Analyzing multilayer foils by confocal Raman microscopy

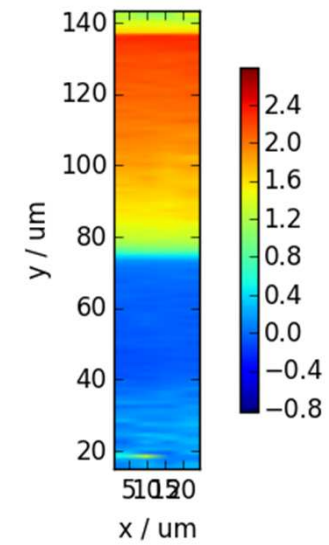
- Resolution $< 1 \mu\text{m}$ in xyz
- Layers are identified by their polymer chemistry
- Layer thickness, thickness variations, distortions, intermixing,... can be mapped
- Adhesives can be identified and their distribution mapped
- No need for microtome sectioning
→ no mechanical artifacts
- Even the morphology of individual layers can be mapped, e.g. semi-crystalline areas



Light microscopy of
microtome cross-section

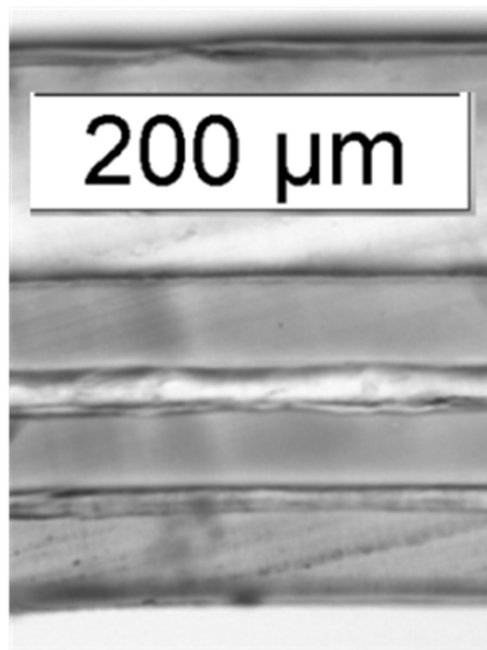


Raman XZ image

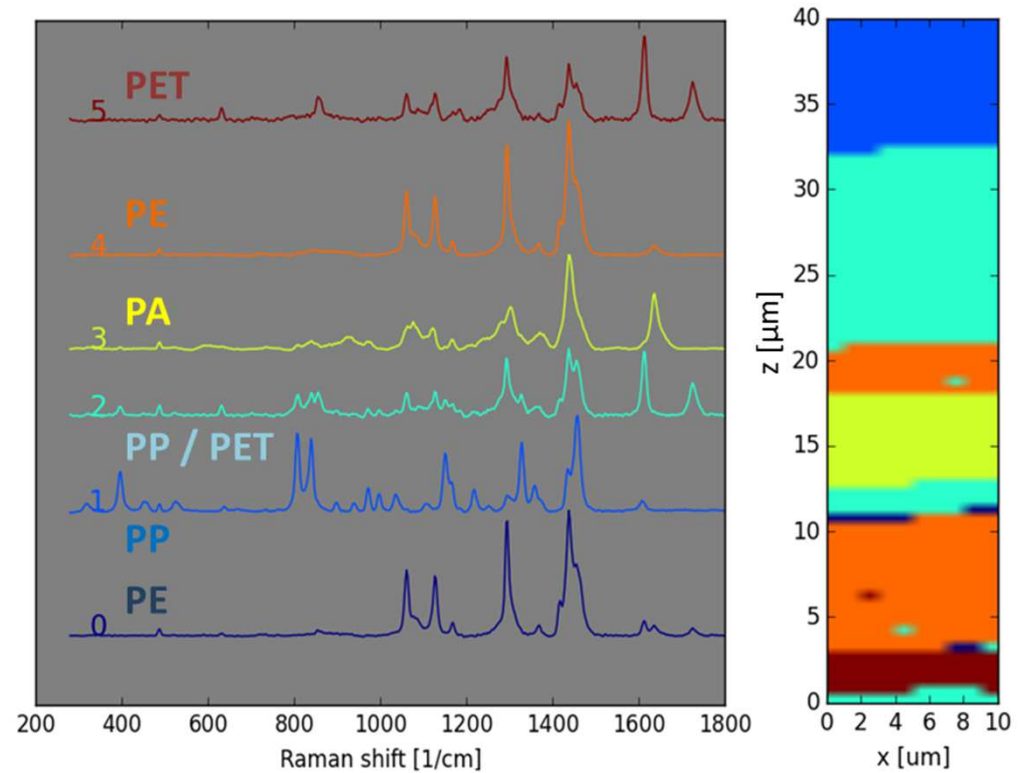


Chemical imaging XZ imaging

of a complex (7-layer) multilayer packaging foil

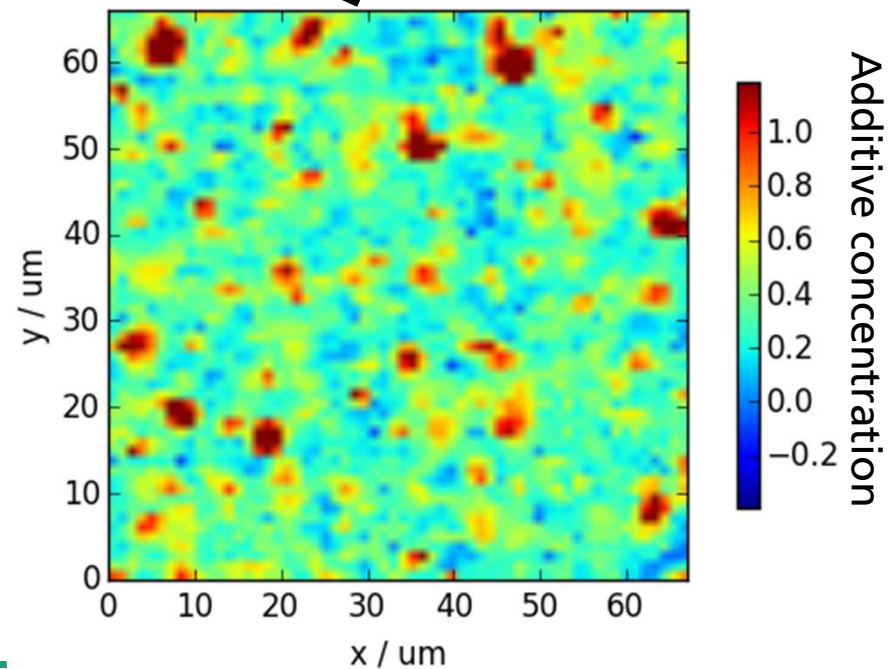
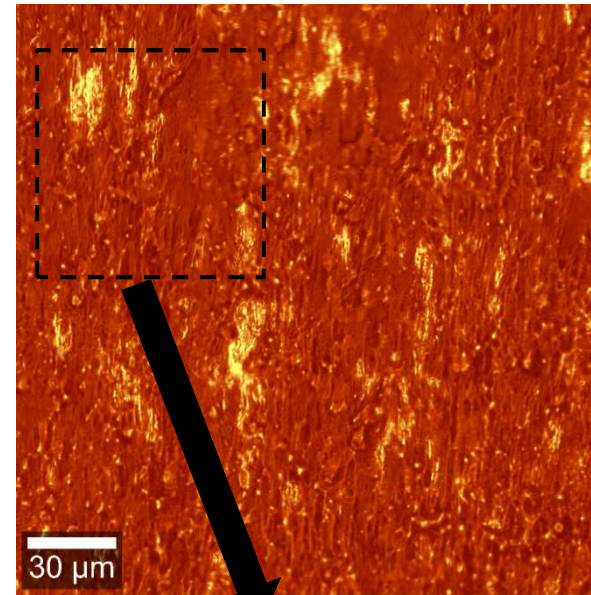
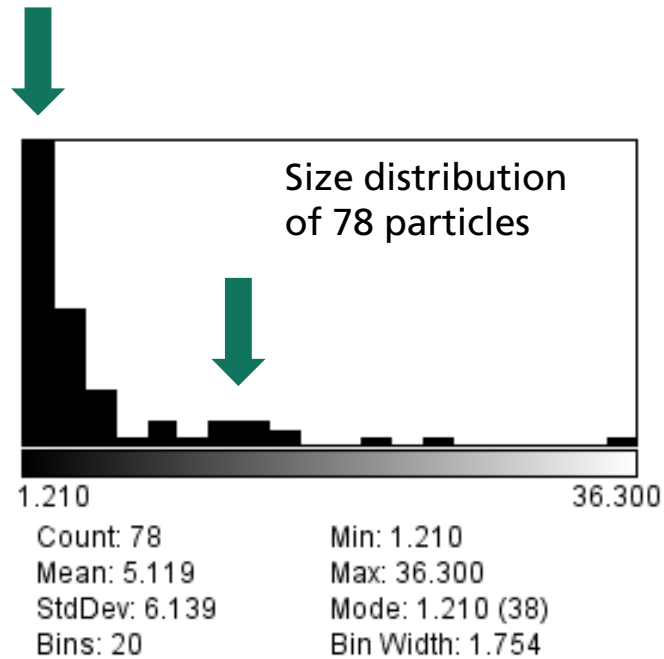


Light microscopy of
microtome cross-section



High resolution imaging of additive distribution within POM

- Approx. 270.000 particles per mm³
- Most particles have diameter < 1 μm
- Some have diameters ~ 5 μm



'ANALYTICS OF STABILIZERS' - INNOVATIVE APPROACHES FOR THE LIQUID CHROMATOGRAPHY ANALYSIS OF ADDITIVES IN POLYMERS

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Forschungsnetzwerk
Mittelstand



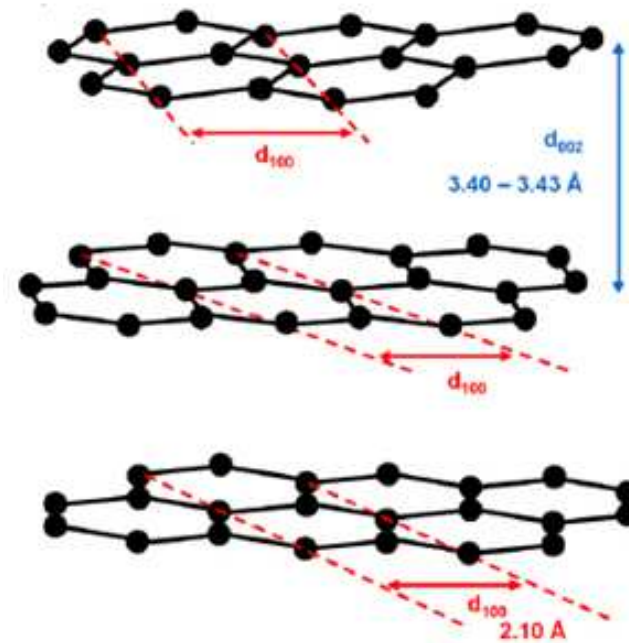
Industrielle
Gemeinschaftsforschung

Additive Analysis

'Generation of an analytical method for the qualitative and quantitative characterization of modern stabilizers and their degradation products in thermoplastic materials'

■ Benefits of this method:

- Separation of additives with high molecular weights and polymers
- Precise quantification
- Analysis of a variety of stabilizers in an one-shot approach



Objective

- I. Separation of polyolefins (PE, PP) and additives (antioxidants, UV absorbers, HALS) using solid phase extraction (SPE)
- II. Qualitative and quantitative characterization of additives and their degradation products – separation of additives and their degradation products using HPLC



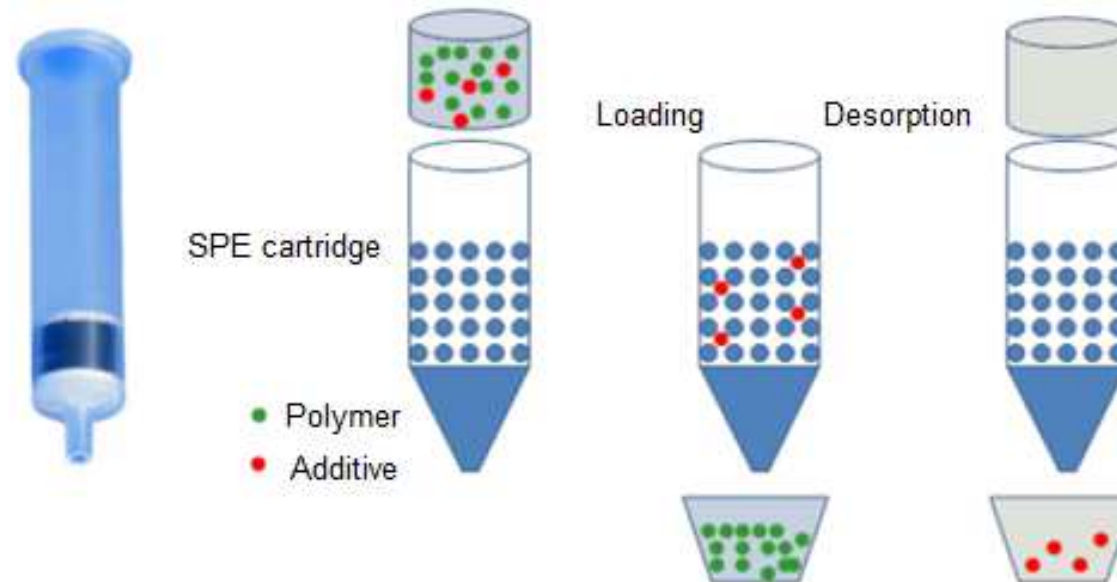
State of the art

- OIT measurements
- IR spectroscopy
- Precipitation of the polymer in solution
- Soxhlet extraction (partly improved extraction, duration of extraction from 6 to 48 h)
- Supercritical Fluid Extraction (SFE)



Realization

- Method: Solid Phase Extraction, SPE



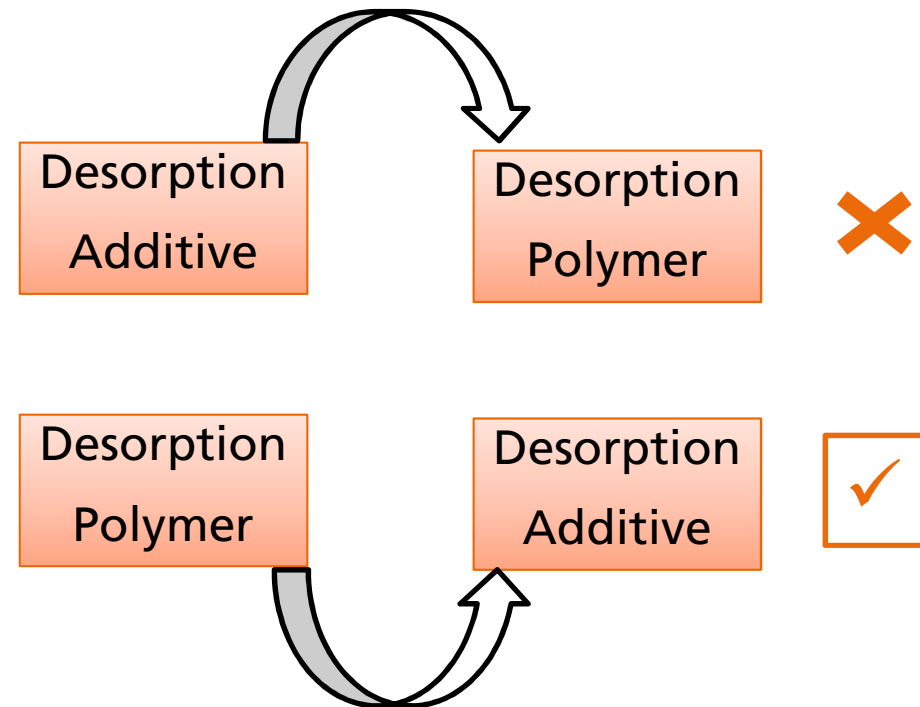
- Separation and analysis of the polymer and additive fractions are performed with the same material as for the HPLC and SPE

Realization

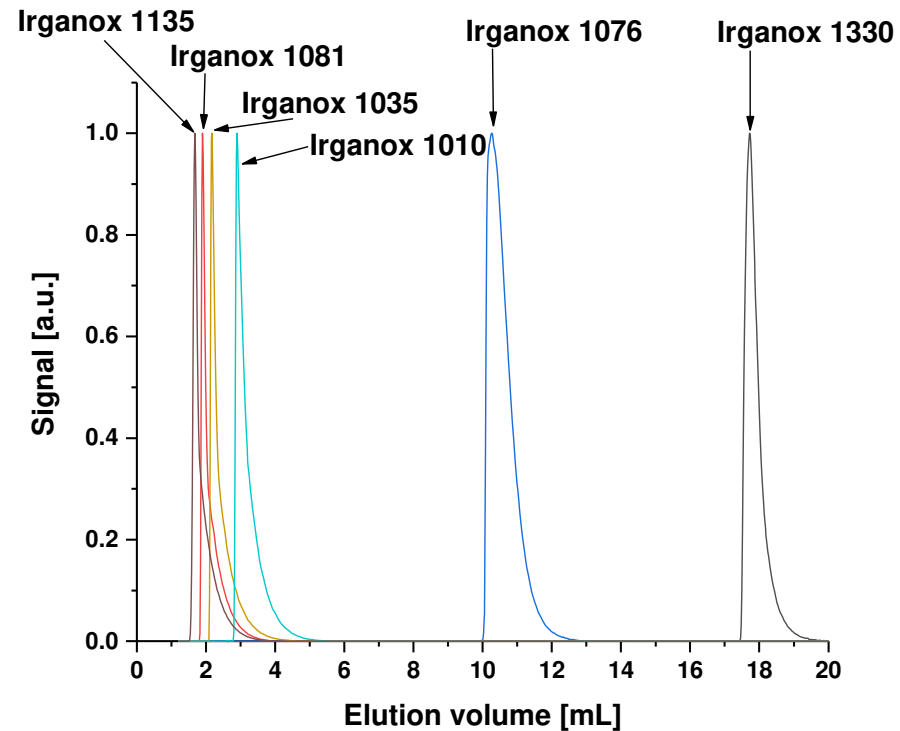
- Oxygen sensitivity of polymer additives
- Approach:
 - Flooding the system with nitrogen or argon
 - Controlling the reproducibility and the extent of the error
- Verification of the received results with a well-known method (e.g. Soxhlet extraction)

Limitation of the method

- Limitation by solubility
- Limitation by adsorption

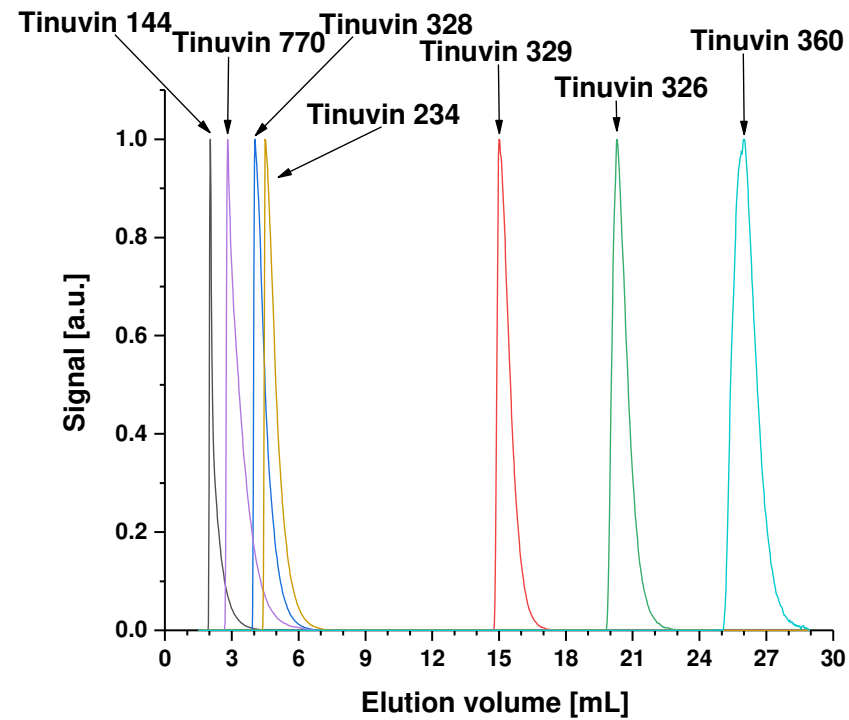


Separation of antioxidants

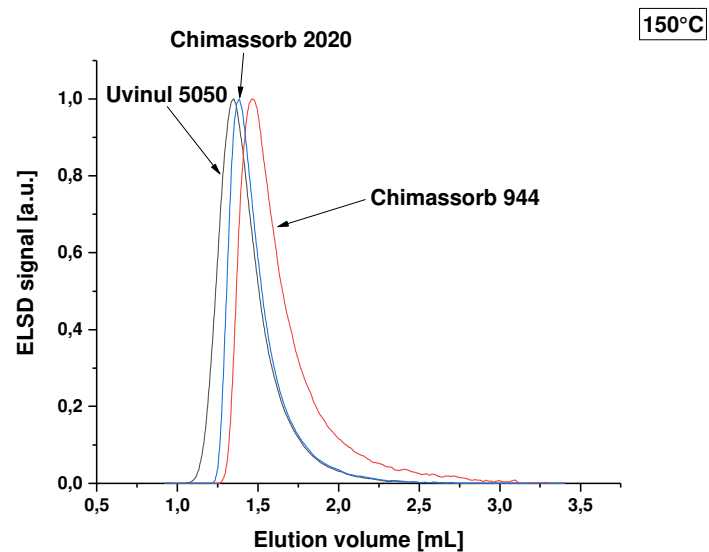
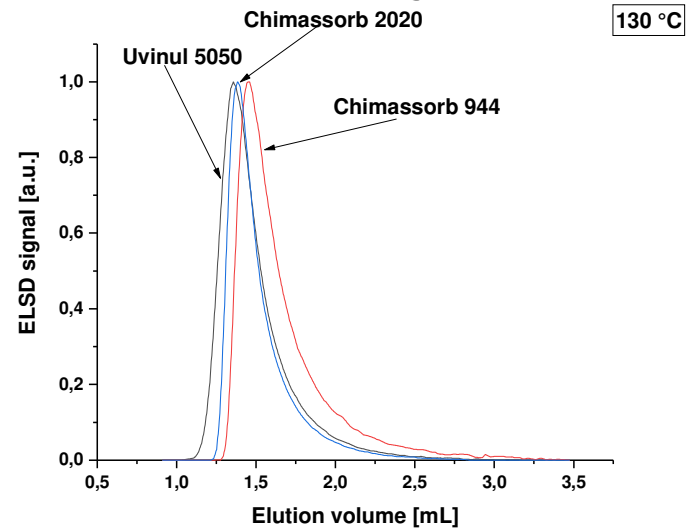
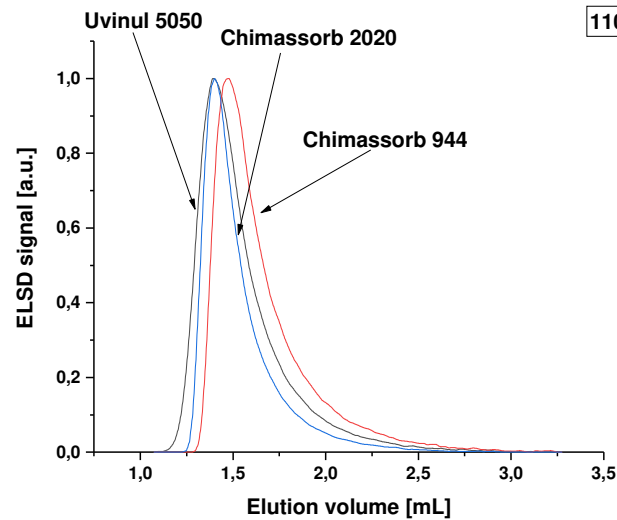


- I. 0 min: 100% MTBE, 0 % CHCl_3
- II. 5 min: 100% MTBE, 0 % CHCl_3
- III. 15 min: 0% MTBE, 100 % CHCl_3
- IV. 30 min: 0% MTBE, 100 % CHCl_3

Separation of UV absorbers

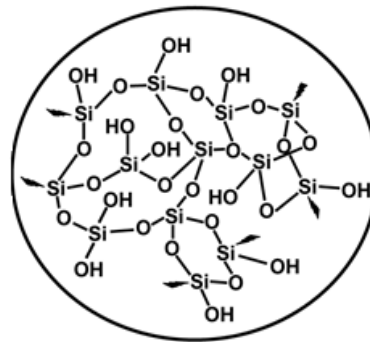
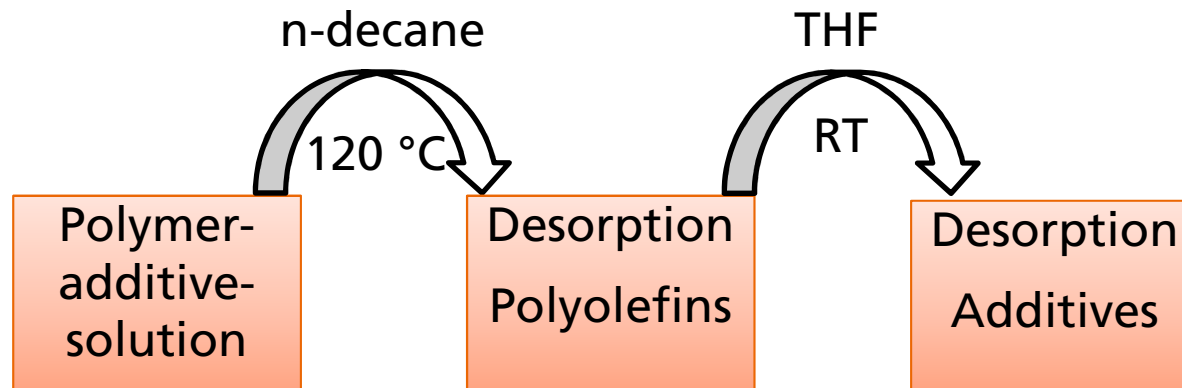


Desorption of oligomeric HA(L)S with mesitylene



- Adsorbent: Toluene
- Desorbent: Mesitylene
- Xylene is a desorbent, but first at 150 °C

Separation of polyolefins and additives using silica as stationary phase



- Result: Separation of polyolefins and additives with silica is possible

Outlook

- Construction of an HT-SPE apparatus
- Separation of polymers and HA(L)S additives
- Quantitative characterization of polymer additives by HPLC